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Free Fatty Acids Esterification on Palm Oil Sludge using Zirconia-supported Indonesian Natural Zeolite as Heterogeneous Catalyst

ARIF HIDAYAT¹* and BACHRUN SUTRISNO¹

¹Chemical Engineering Department, Universitas Islam Indonesia, Yogyakarta 55584, Indonesia. *Corresponding author E-mail: arif.hidayat@uii.ac.id

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ABSTRACT

Biodiesel production has been attracted more intense study because of the growing interest in finding the alternative renewable energy resources. Commonly biodiesel production is produced using refined vegetable oils as feedstocks which catalyzed by a homogeneous catalyst. However, the homogeneous catalyst has many disadvantages such as complex products separation and environmentally problem. In this research, the esterification of Palm Oil Sludge (POS) using zirconia supported Indonesian natural zeolite as a heterogeneous catalyst was studied. The POS is low feedstock price and available abundantly. The catalysts were analyzed in term of porosity analysis (Gas Sorption Analysis method), XRD and FTIR. The FFAs conversion as high as was 84.2% achieved at following reaction conditions: 2 h reaction time, 65°C reaction temperature, 1:10 POS to methanol mole ratio, and catalyst amount 10 wt.% of POS. The FFA conversion was decreased after three consecutive of esterification reaction.

> Keywords: Esterification, Sludge palm oil, Biodiesel, Natural zeolite, Free fatty acids, Sulfated zirconia

INTRODUCTION

The biodiesel development as the sources of alternative renewable energy to substitute diesel fuel from crude oil has gained more attention in the last few decades due to the decreasing of fossil fuel reserves and environmental concern. It has several benefits such as biodegradable, environment friendly, and less toxic. Biodiesel can be generated from animal fat or vegetable oil through a transesterification reaction. Conventionally, a base

catalyst, such as potassium and sodium hydroxides, is used to catalyze the transesterfication in biodiesel production. The transesterification esterification reactions are expressed in Figure 1.

Many types of vegetable oil sources such as soybean¹, corn^{2,3}, palm oil^{4,5}, coconut⁶, sunflower⁷, canola8, and bran9 can be utilized as feedstocks to produce biodiesel. However, the utilization of vegetable oil as a feedstock for biodiesel production



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has many drawbacks i.e. less economic prospect and impractical due to high feedstock price and competition as a food supply¹⁰. To solve the problem, low-grade oil or non-edible oil which has low price can be used as biodiesel feedstocks¹¹. This type of feedstocks contains high free fatty acids (FFAs) so that lead to saponification when using base catalysts. To overcome the problem, the acid catalysts should be applied for converting FFAs to biodiesel when using these feedstocks.

$\mathrm{CH}_2\mathrm{COOR}_1$			base	R1COOCR	CH_2OH
CHCOOR ₂	+	3 ROH	$\stackrel{catalyst}{\Leftrightarrow}$	R ₂ COOCR +	 C H OH
 CH2COOR3 Triglycerides		Alcol	hal	R3COOCR Biodiesel	 CH2OH glycerol
		_			

Fig. 1. Transesterification reactions

In 2017, the palm oil industries in Indonesia has generated 2,4 million tons of palm oil which means generating low-grade oil in large amounts such as palm oil sludge (POS) as wastes¹². The POS has a large amount of free fatty acids (FFAs) that be reacted with an alcohol using acid catalyst to form biodiesel. Utilization of POS can be replaced the refined vegetable oils feedstocks in order to enhance the production of biodiesel potential economically.

Biodiesel can be synthesized by esterification reactions of FFAs on POS with an alcohol to produce methyl esters and water. Sulfuric acid, as well as hydrochloric acid or nitric acid, are commonly employed as homogeneous catalysts to accelerate the reaction. Fig. 2. is expressed the esterification reactions for producing esters or biodiesel.

		acid		
R ₂ CHOOH	+ ROH	\Leftrightarrow	R ₁ COOR +	H_2O
Free fatty acid	Alcohol		Biodiesel	water

Fig. 2. Esterification reactions

The esterification reactions have many limitations, i.e. requires advances process for purification of products, further separation of catalyst, and furthermore the homogeneous catalysts are not reusable¹³. In order to eliminate the limitations related to using homogeneous catalysts, the heterogeneous catalyst is applied in biodiesel production. The heterogeneous catalysts offer several benefits such as produce fewer amounts of chemical waste and wastewater, not necessary advances process separation, has a hydrophobic surface, and less corrosion. Many studies have been reported the using of heterogeneous catalyst for biodiesel production, such as sulphated zirconia $(SO_4^{2-}/ZrO_2)^{14-16}$, zeolite^{17, 18,19,20}, Lanthanum (La³⁺) and HZSM-5²¹, Natural clay²²⁻²⁴, Amberlyst 15²⁵, alumina (Al₂O₃)²⁶, silica²⁷, hydroxyapatite²⁸, Tungsten supported titania-silica composite²⁹, Tin oxide³⁰, niobic³¹, SBA-15^{32,33}, Titania³⁴ and biomass waste biomass wastes^{10,12, 35-37}.

However, most of the heterogeneous catalysts as reported have complex preparation steps and expensive synthesis routes. The heterogeneous catalysts should be synthesized by simple preparation steps, available abundantly and low in price to reduce the catalyst production cost. Zeolite is an aluminosilicate natural mineral which has a crystalline framework three-dimensional structure to form pores with uniform size. Zeolite has been applied in many industrial applications, such as adsorbent in wastewater treatment, molecular sieve, catalyst, etc. The study related to the Indonesia natural zeolite application for biodiesel production as a solid catalyst is still limited. The esterification reactions of FFAs with methanol, using zirconia supported on Indonesia natural zeolite, were an investigation in this research. The natural zeolite catalyst activity to convert FFAs to biodiesel was tested in terms of the reaction temperatures, POS to methanol mole ratio, catalyst amounts and reaction time. The catalyst reusability for consecutive reactions was also investigated in this work.

EXPERIMENTAL

Preparation of Catalyst

Prior to use, natural zeolite collected from Gunung Kidul province, Indonesia was crushed to obtain small particle with a size of 200 mesh. Then, to remove the organic impurities from the internal pores, the zeolite was activated by mixing with sulfuric acid 0.5 N solutions. The activation was performed by heating the solution containing natural zeolite at 80°C for 4 h while vigorous stirring. Subsequently, the activated natural zeolite was separated from solutions by filtration and rinsed with distilled water until the wash water free from sulfate ions. Then, the activated natural zeolite was dried in an oven to remove remaining water from internal pores for overnight at 110°C. The catalyst was prepared by impregnation of Zirconia metals on the activated natural zeolite. The catalyst synthesis procedure is described as follows: (i) 10 gr of Zirconium (IV) oxychloride octahydrate (ZrOCl₂.8H₂O) was added into 100 mL distilled water and mixed well, (ii) followed by the addition of activated natural zeolite into the ZrOCl₂.8H₂O salt solutions, (iii) the mixture was agitated at 60°C for 8 h, after completing the impregnation procedure, the mixture was dried to remove remaining water in an oven at 130°C for 8 hours. Finally, the sample was calcined in a furnace at 550°C for 3 hours.

The BET (Brunauer–Emmet–Teller) method was applied to determine specific surface area of catalysts. BET method was based on the physical adsorption of nitrogen between relative pressures (P/P0) of 0.05–0.35 at 77 K. The surface functional groups were characterized using Fourier Transform Infrared (FT-IR) spectroscopy which performed in the wavelength number between ranges of 400-4000 cm⁻¹. The materials crystalline structures were determined using X-Ray Diffraction at operating condition 40 kV and 40 mA with 20 measurements between the ranges of 5° to 80°.

Activity Test

Esterification of FFA on POS was run in a 250 mL two-neck flat-bottom flask as a batch reactor. A hot plate equipped a magnetic stirrer was used to stir the reactant at the batch reactor. The batch reactor was connected to a condenser and equip with a thermometer to indicate reaction temperature. Firstly, a certain amount of catalyst was mixed to a known volume of methanol and raised to the selected temperature under continuous stirring. Subsequently, POS was poured slowly into the reactor. The reaction was conducted till the completion of the reaction (2 h reaction time). The reaction conditions were varied in terms of POS to methanol mole ratio, catalyst amount, and reaction temperature to obtain optimum conditions. When the reaction was completed, the biodiesel product was separated from the catalyst by filtration. The product mixture was put into a separating funnel and kept at room temperature for overnight, after which two liquid phases appeared. The upper layer consists of the biodiesel product, and whiles the remaining methanol and water were in the bottom. The acid value (AV) of the upper layer was measured by titration method. The AV is described as the weight of potassium hydroxide in mg required to neutralize 1 g of oil contains of the FFAs. Initially, the samples were heated in the oven at 110°C in order to eliminate the remaining of methanol and water. The titration technique was carried out by using 0.1 M KOH solution as the titrant, while phenolphthalein as indicator. The volume of alkaline solution consumed was noted, and the AV of sample was determined using Equation 1.

$$AV = \frac{V.0.1.56.1}{m}$$
(1)

where AV is the sample acid value (mg KOH/g), V is the KOH solution volume used in the titration (mL), 0.1 is the KOH solution concentration (mol/L), m is the sample weight (g), and 56.1 is the KOH molecular weight. The FFAs conversion (X_F) was calculated using Equation 2.

$$X_F(\%) = \left(1 - \frac{acid value after reaction}{initial acid value}\right) x \ 100 \tag{2}$$

The catalyst reusability test was performed under the same optimum reaction conditions in the experiment with the fresh catalyst. The reused catalysts were washed with n-hexane and dried in an oven at 100°C for 4 h prior to reuse. Three consecutive of esterification reactions were carried out to evaluate the catalyst reusability.

RESULTS AND DISCUSSION

Catalyst Characterization

The crystalline structure of activated natural zeolite and zirconia catalyst were presented in Fig. 3 (a) and (b). As presented in Fig. 3 (a), the activated natural zeolite diffraction peaks shows the characteristics of mordenite and clipnotile zeolites which appears at $2^{\circ}\theta = 9.76$ (2 0 0), 19.68 (4 0 0), 22.38 (1 5 0), 25.78 (2 0 2), 27.68 (5 1 1), and 30.05 (1 5 1). In the other hand, the catalyst presents characteristic peaks of zirconia oxide at $2^{\circ}\theta = 30.47 (2 \ 2 \ 1), \ 30.87 (1 \ 1 \ 1) \ and \ 50.55 (2 \ 2 \ 0).$ It can be observed that the catalysts retained the ordered structure. The intensity decreasing could be attributed the loading of zirconia on the solid surface. The shifting of diffraction peaks corresponds to the cell parameters due to the shrinkage of pores framework after the impregnation of zirconia.

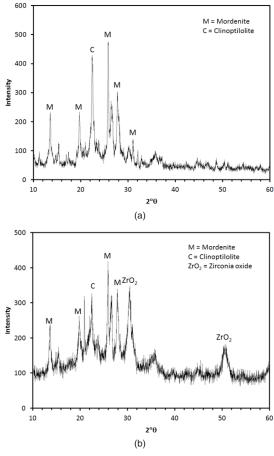


Fig. 3. The XRD characteristics of activated natural zeolite and zirconia catalyst

The activated natural zeolite and zirconia catalyst N_2 adsorption–desorption isotherms were exhibited in Fig. 4 As presented in Fig. 4, the activated natural zeolite as well as zirconia catalyst exhibited the type IV of isotherms with H4 hysteresis loops which associated with the characteristic of mesoporous materials with narrow slit-like pores, internal voids of irregular shape while the pore size distribution is broad. A decrease in the volume of nitrogen adsorption indicates a reduction in surface area after zirconia impregnation on the activated natural zeolite.

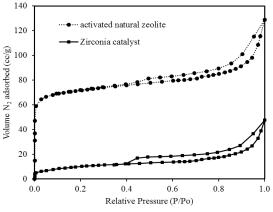


Fig. 4. The activated natural zeolite and zirconia catalyst $\rm N_{2}$ adsorption–desorption isotherms

The BET surface area, pore volume, and average pore size diameter of the activated natural zeolite and zirconia catalyst are shown in Table 1. The activation process was successful in increasing of the natural zeolite surface area from 177.542 to 213.421 m².g⁻¹. The pore volume also increased from 0.2005 to 0. 0.2107 cm.g⁻¹. The increasing of surface area and pore volume is due to the removal of impurities on the natural zeolite internal pores by activation process using chemical treatment. Furthermore, the impregnation of zirconia metal as active sites would reduce the surface area to 115.348 m².g⁻¹ as well as pore volume (0.07892 cm.g⁻¹). The average pore diameter widened from 3.74 to 8.28 nm after impregnation of Zirconia metal on the surface of the catalyst. An increase in mean pore diameter will facilitate accessing of free fatty acid compounds to penetrate into the internal pores of catalyst materials.

The successful of incorporated of zirconia metals on catalyst surface was determined using Energy Dispersive X-ray Spectroscopy (EDS) to calculate elemental composition. Before impregnation step, the activated natural consist of Na (0.62% w/w), Al (6.13% w/w) and Si (27.83% w/w) respectively. While after impregnation, the zirconia catalyst

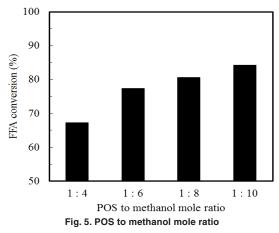
Sample	BET surface area (m² g⁻¹)	Average pore size diameter (nm)	total pore volume (cm ^{3 g-1})
Natural zeolite	177.542	2.26	0.2005
Activated natural zeolite	213.421	3.74	0.2107
Zirconia catalyst	115.348	8.28	0.07892

Table 1: Physicochemical characterics of samples

containing Na (6.13% w/w), Al (6.13% w/w), Si (6.13% w/w) and Zr (6.13% w/w). The FT-IR spectra for both activated natural zeolite and zirconia catalyst were presented in Fig. 4. In natural zeolite FT-IR spectra, the wavelength between 2600 and 3700 cm⁻¹ corresponds to the OH bond stretching vibration of H₂O is clearly visible. The peak between 1000 and 1300 cm⁻¹ indicate the asymmetric Si-O-Si. The Si-O-Si bending bond occurs at a wavelength of 458 cm⁻¹. Meanwhile, the peak of 757 and 811 cm⁻¹ shows the symmetric vibration of Si-O-Si bonds. In the zirconia catalyst spectra, the OH stretching vibration of water was at 3426 cm⁻¹. The wavelength number at 1090 and 1216 cm⁻¹ correspond to asymmetric Si–O–Si vibrations which are seen to reduced in intensity compared to activated natural zeolite. The appearance of the bands between from The T–O–T bonds (T = Si or Al) appears at 1088 and 1230 cm⁻¹ which correspond to Si-OH vibrations. The intensity of this band tends to increase when several metals are incorporated of the metals into the surface of materials.

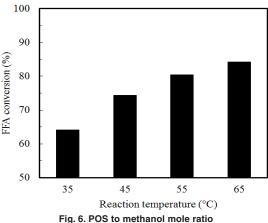
Activity Test Mole Ratio Effect

The important parameter that affects the conversion of FFAs is the oil to methanol mole ratio. Mole ratio is the ratio between the mole number of oil to the mole number of alcohol that used in the experiment. Theoretically, the FFAs esterification reaction requires one mole of alcohol for each mole of oil that reacted. However, in practice, the mole ratio must be provided more than the stoichiometric ratio requirement to drive the equilibrium to the methyl ester product. The POS to methanol mole ratio is varied at a number value of 1:4, 1:6, 1:8, and 1:10, while the reaction was kept constant at the identical operating conditions (reaction temperature of 60°C, catalyst amount of 5%wt. of oil, dan reaction time of 4 h). It was concluded that an enhancement of methanol mole number leads to an increasing of conversion of FFAs as shown in Fig. 5. Initially, the FFA conversion was obtained at 67.2% as the POS to methanol mole ratio was 1:4. Furthermore, the FFA conversion increased from 77.3% to 80.7% at the POS to methanol mole ratio of 1:6 to 1:8 molar ratio respectively. Furthermore, the FFA conversion increases to 84.22% in the oil to methanol mole ratio of 1:10.



Reaction Temperature Effect

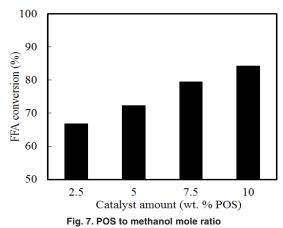
The reaction temperature effect on the conversion of FFAs was investigated by conducting esterification reactions at different temperatures (35, 45, 55, and 65°C), while the POS to methanol mole ratio and catalyst amount was maintained constantly at 1:10 and 10 wt. % of POS, respectively. Fig. 6 exhibits the conversion of FFAs at different reaction temperatures. As illustrated in Fig. 6, the results indicated that the reaction temperatures had an effect on FFA conversion during the esterification reaction. The FFA conversion achieves at 64.2% after time reaction of 2 h and reaction temperature of 35°C. Subsequently, the FFA conversion reached to 74.4 and 80.4% at reaction temperature of 45 and 55°C, respectfully. Noteworthy, the maximum FFA conversion acquired at 84.2% when the reaction temperature of 65°C. It was can be summarized that the FFA conversion on the product increased with the rising of reaction temperatures. The high of FFAs conversion was due to the acceleration of reaction rate which caused more methyl ester produced. The solubility of the oil in methanol will increase at higher temperatures, thus the problems associated with mass transfer limits will be reduced.



Catalyst Amount Effect

The catalyst amount is the important parameter in the biodiesel production using a heterogeneous catalyst. The heterogeneous catalysts preferably are employed due to easy separation from product mixture and regenerable. The appropriate catalyst amount to achieve maximum FFA conversion is investigated by varying in the range of 2.5, 5, 7.5 and 10 wt.% of POS. The percentage values were the mass fractions of the POS applied to the reactant mixture. The experiments were conducted with a reaction temperature of 60°C and a fixed POS to methanol mole ratio of 1:10, respectively. The conversion of FFAs at different reaction temperatures was presented in Figure 7.

Increasing the catalyst amount from 2.5 to 5 wt.% of POS increased the FFA conversion from 66.9 to 72.3%. Among the catalyst amount variations studied, the highest FFA conversion was 84.2% at the catalyst amount of 10 wt.% of POS. The results exhibited the improvement of FFA conversion when the catalyst amount was increased. The increase in FFA conversion is due to the addition of catalysts amount will lead the more of active sites available in catalyst surface. Hence, the optimum amount for FFA conversion is considered at the catalyst amount of 10 wt.% of POS.

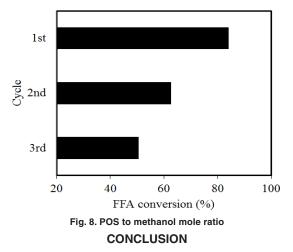


Catalyst Reusability

The catalyst performance associated with repeated use is investigated by subsequent reaction cycles. Prior to reuse, the catalysts were activated

1. Bhandari, R.; Volli, V.; Purkait, M. K., *J. Environ. Chem. Eng.*, **2015**, *3*, 906-914.

by washing them with hexane to remove residual oil and impurities on the catalyst surface. After that, the catalysts were further dried in the oven at 120°C in the oven for 4 hours. Then the catalysts were applied for an esterification reaction under the identical reaction conditions as before. The result of reusability test is depicted in Fig. 8. As shown in Fig. 8, the FFA conversion was reduced in the second and third consecutive reactions.



The biodiesel production from Palm oil sludge (POS) containing high free fatty acids (FFA) using Zirconia natural zeolite catalyst was studied. The catalyst was prepared by impregnation of Zirconia metal on the activated natural zeolite surface. The physico-chemical characteristics of catalysts were analyzed in term of porosity analysis, XRD, elemental composition and FTIR. The optimum operation conditions of esterification reaction was POS to methanol mole ratio of 1:10, catalyst amount of 10 wt.% POS, and reaction temperature of 65°C. At this optimum condition, the FFA conversion achieves 84.2%.

ACKNOWLEDGEMENT

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REFERENCES

2.

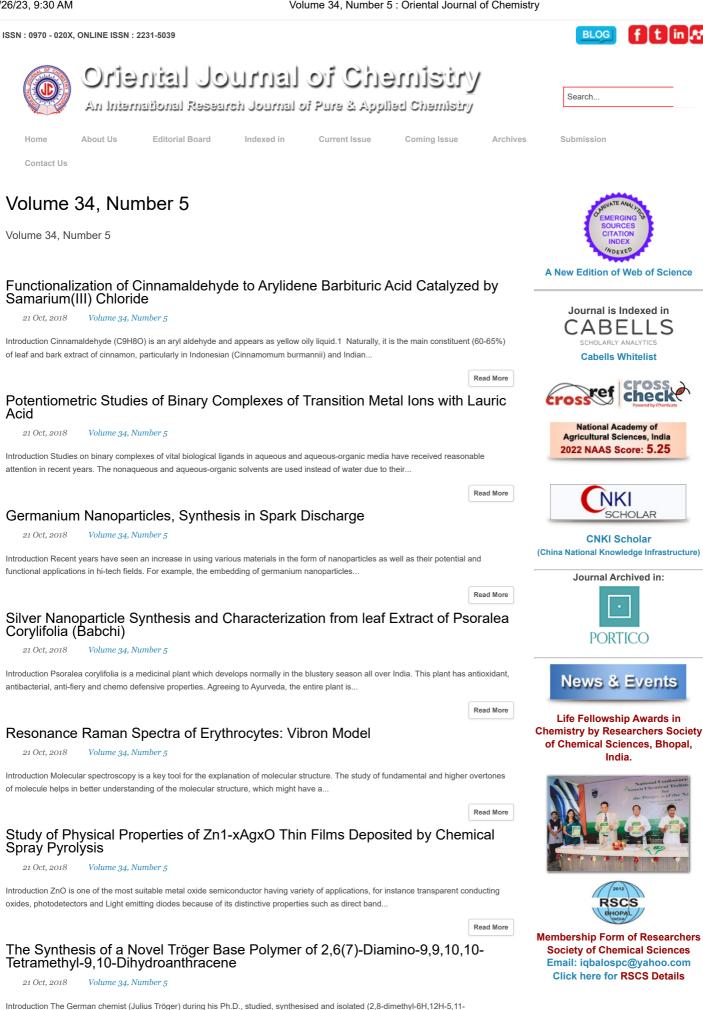
Sirajuddin, M.; Tariq, M.; Ali, S., *J. Organomet. Chem.*, **2015**, *779*, 30-38.

- Abo-Dief, H. M.; Emam, A. S.; Abualnaja; K. M.; Mohamed, A. T., *Orient. J. Chem.*, 2018, 34, 1011-1015.
- Kadapure, S. A.; Kirti, P.; Singh, S.; Kokatnur, S.; Hiremath, N.; Variar, A.; Shaikh, S.; Chittaragi, R., *Int. J. Sustain. Eng.*, **2017**, 1-6.
- Nurcahyo, I. F.; Wijaya, K.; Triyono; Budiman, A.; Taufiq-Yap, Y. H., *Orient. J. Chem.*, **2018**, *34*, 750-756.
- Qiu, T.; Guo, X.; Yang, J.; Zhou, L.; Li, L.; Wang, H.; Niu, Y., *Chem. Eng. J.*, **2016**, *296*, 71-78.
- Feyzi, M.; Hosseini, N.; Yaghobi, N.; Ezzati, R., *Chem. Phys. Lett.*, **2017**, *677*, 19-29.
- Maleki H.; Kazemeini M.; Bastan F., *Chem. Eng. Technol.*, **2017**, *40*, 1923-1930.
- Hidayat, A.; Mukti, N. I. F.; Handoko, B.; Sutrisno, B., *Int. J. Tech.*, **2018**, *9*, 400-411.
- Augustia, V. A. S.; Djalal, R. A.; Sutrisno, B.; Hidayat, A., *IOP Conf. Ser. Mater. Sci. Eng.*, **2018**, *105*(1),012065.
- 11. Hidayat, A.; Sutrisno, B., *AIP Conf. Proc.*, **2017**, *1803*, 020046.
- 12. Hidayat, A.; Sutrisno, B., *AIP Conf. Proc.*, **2017**, *1840*, 050001.
- Pradana, Y. S.; Hidayat, A.; Prasetya, A.; Budiman, A., *Defect Diffus. Forum.*, **2018**, *382* DDF, 280-285.
- 14. Nguyen, H. K. D.; Van Pham, P.; Vo, A. D., *J. Porous Mat.*, **2017**, *24*, 411-419.
- Raia, R. Z.; da Silva, L. S.; Marcucci, S. M. P.; Arroyo, P. A., *Catal. Today.*, **2017**, *289*, 105-114.
- Saravanan, K.; Tyagi, B.; Shukla, R. S.; Bajaj, H. C., *Appl. Catal. B Environ.*, **2015**, *172-173*, 108-115.
- 17. Prinsen, P.; Luque, R.; González-Arellano, C., Micropor. Mesopor. Mat., 2018, 262, 133-139.
- Baroi, C.; Dalai, A. K., *Appl. Catal. A Gen.*, 2014, *485*, 99-107.
- 19. Doyle, A. M.; Albayati, T. M.; Abbas, A. S.; Alismaeel, Z. T., *Renew. Energ.*, **2016**, *97*, 19-23.
- 20. Purova, R.; Narasimharao, K.; Ahmed, N. S. I.; Al-Thabaiti, S.; Al-Shehri, A.; Mokhtar, M.;

Schwieger, W., *J. Mol. Catal. A Chem.*, **2015**, *406*, 159-167.

- Vieira, S. S.; Magriotis, Z. M.; Santos, N. A. V.; Saczk, A. A.; Hori, C. E.; Arroyo, P. A., *Bioresour. Technol.*, **2013**, *133*, 248-255.
- 22. Rezende, M. J. C.; Pinto, A. C., *Renew. Energ.*, **2016**, *92*, 171-177.
- 23. Aghabarari, B.; Dorostkar, N., *J. Taiwan Inst. Chem. Eng.*, **2014**, *45*, 1468-1473.
- Kanda, L. R. S.; Corazza, M. L.; Zatta, L.; Wypych, F., *Fuel.*, **2017**, *193*, 265-274.
- Hykkerud, A.; Marchetti, J. M., *Biomass Bioenerg.*, **2016**, *95*, 340-343.
- Ma, G.; Hu, W.; Pei, H.; Jiang, L.; Ji, Y.; Mu R., Environ. Technol. (UK)., 2016, 36, 622-627.
- Pandiangan, K. D.; Simanjuntak, W.; Rilyanti, M.; Jamarun, N.; Arief, S., *Orient. J. Chem.*, 2017, *33*, 2891-2898.
- Yan, B.; Zhang, Y.; Chen, G.; Shan, R.; Ma, W.; Liu, C., *Energ. Convers. Manage.*, **2016**, *130*, 156-164.
- Kaur, M.; Malhotra, R.; Ali, A., *Renew. Energ.*, 2018, *116*, 109-119.
- Nuithitikul, K.; Prasitturattanachai, W.; Hasin, W., Energy Procedia., 2017, 138, 75-80.
- Mohammed, N. I.; Kabbashi, N. A.; Alam, M. Z.; Mirghani, M. E. S., *J. Taiwan Inst. Chem. Eng.*, **2016**, *63*, 243-249.
- 32. Patel, A.; Brahmkhatri, V.; *Fuel Process. Technol.*, **2013**, *113*, 141-149.
- Castanheiro, J. E.; Fonseca, I. M.; Ramos, A. M.; Vital, *J., Micropor. Mesopor. Mat.*, **2017**, *249*, 16-24.
- 34. Essamlali, Y.; Larzek, M.; Essaid, B.; Zahouily, M., Ind. Eng. Chem. Res., **2017**, *56*, 5821-5832.
- Pradana, Y. S.; Hidayat, A.; Prasetya, A.; Budiman,
 A., *Energy Procedia.*, **2017**, *143*, 742-747.
- Hidayat, A.; Rochmadi; Wijaya, K.; Budiman,
 A., *AIP Conf. Proc.*, **2015**, *1699*, 050005.
- Hidayat, A.; Rochmadi; Wijaya, K.; Budiman, A., *IOP Conf. Ser. Mater. Sci. Eng.*, **2016**, *105* (1), 012026.





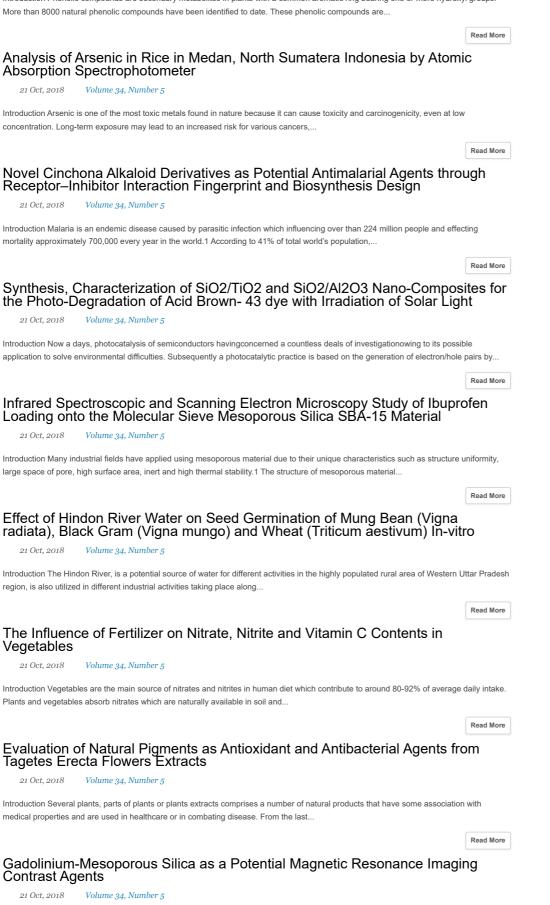
toluidine) in an acid-catalysed media.1 Through ...

methanodibenzo[b,f][1,5]diazocine) also known as Tröger base 1 from the condensation of methanal (formaldehyde) with 4-aminotoluene (p-

Structure Activity Relationship Analysis of Antioxidant Activity of Simple Benzene Carboxylic Acids Group Based on Multiple Linear Regression

21 Oct. 2018 Volume 34, Number 5

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Introduction A novel class of anti-diabetic drugs, which are inhibitors of dipeptidyl-peptidase IV (DPP4), which included sitagliptin, vildagliptin and saxagliptin.1,2,3,4,5 Type 2 diabetes mellitus (T2DM) is a progressive disease, for the treatment of many.

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Utilization of Waste Leaves Biomass of Myrica Esculenta for the Removal of Pb (II), Cd (II) and Zn (II) lons from Waste Waters

21 Oct. 2018 Volume 34, Number 5

Introduction Since from last two or three decades, the exposure of heavy metals in the aquatic bodies has increased very rapidly.1 The presence of heavy metals in the water or waste water causes life threatening.

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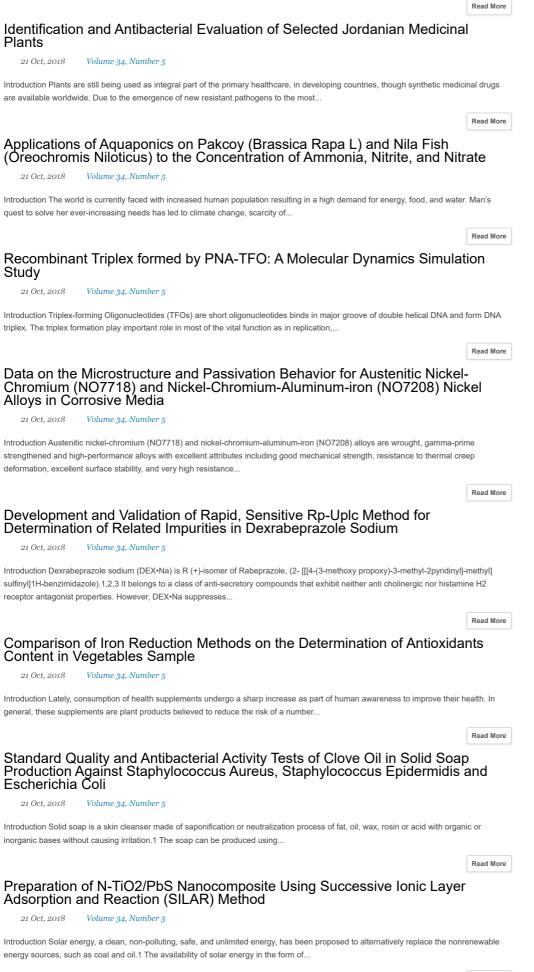
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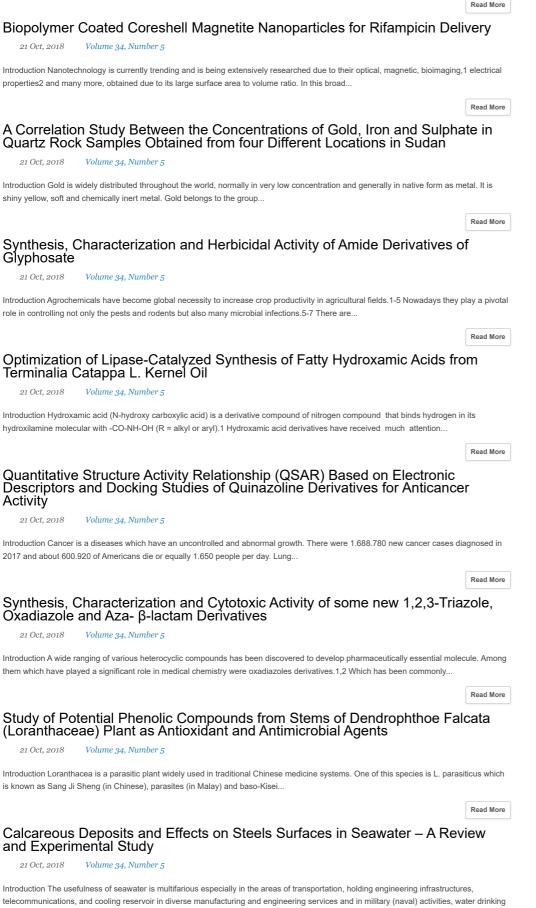
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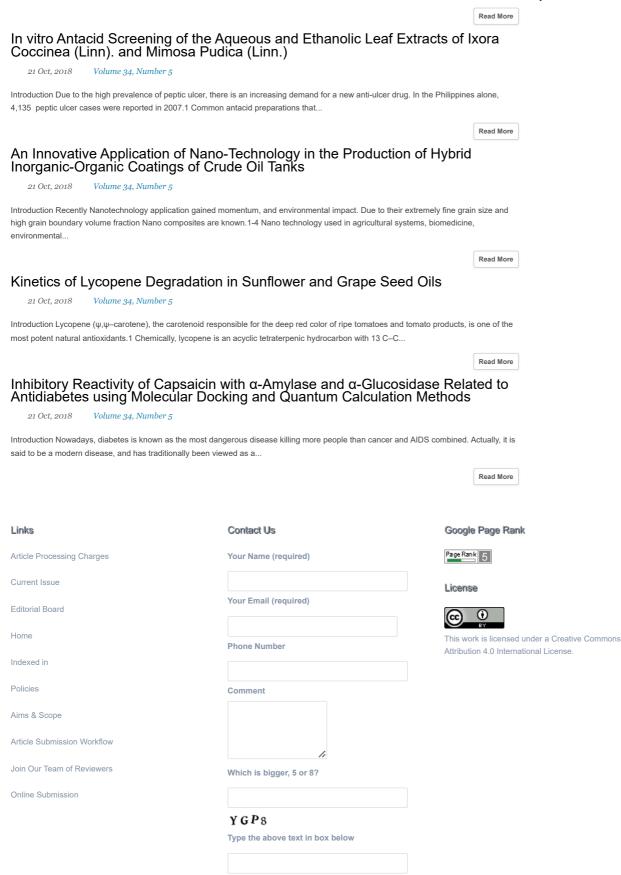


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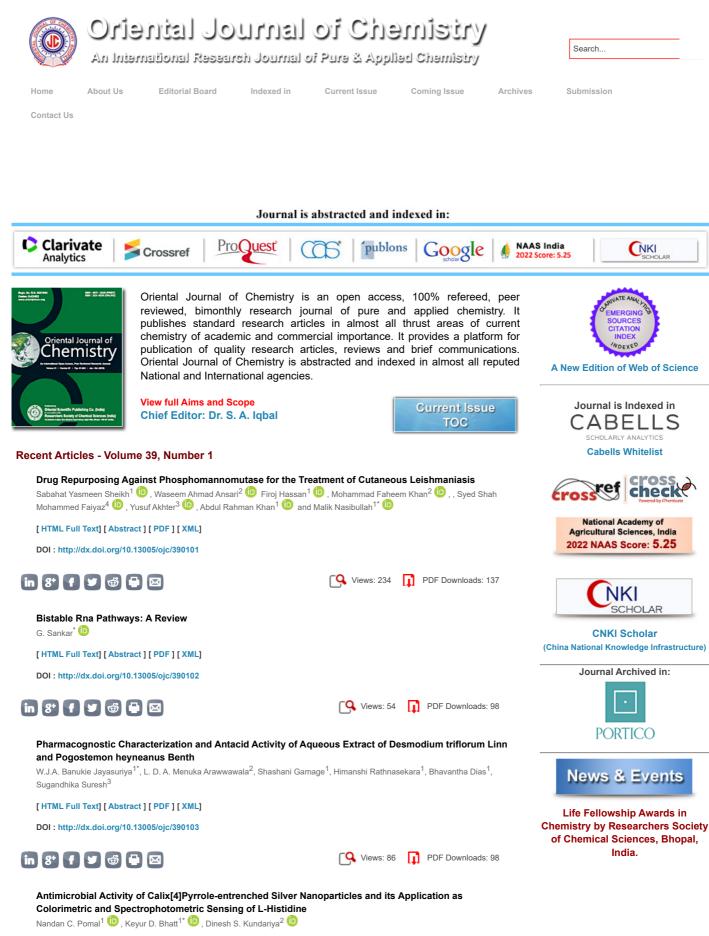
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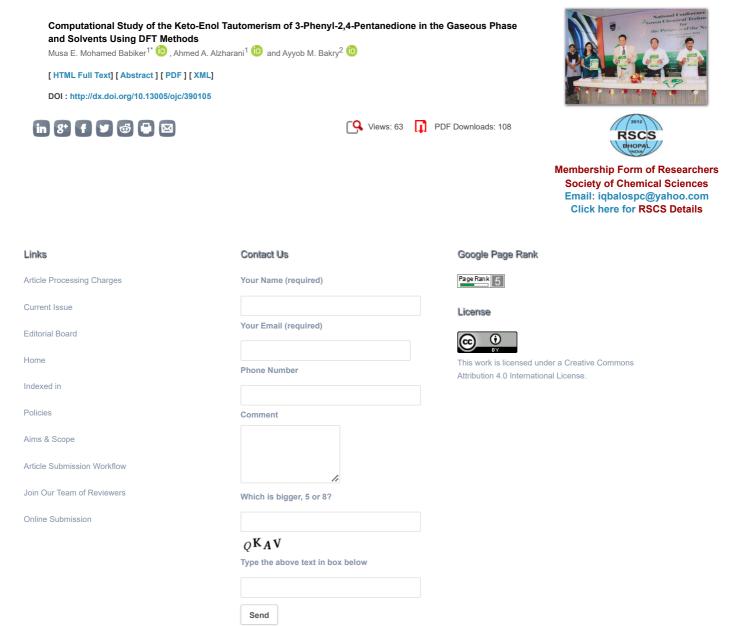
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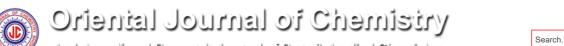
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Prof. Mushahid Husain Jamia Millia Islamia New Delhi- 110 025 India mhusain1@imi.ac.in

Scopus ID: 7201522764

Dr. Nenad Ignjatović Institute of Technical Sciences Serbian Academy of Science and Arts, Knez Mihailova 35/4 Belgrade, 11000, Serbia dr.nenad.ignjatovic@gmail.com Scopus ID: 6602122010

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Guemra_2@yahoo.fr

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ojolosunday@yahoo.com Scopus ID: 25648236700

Dr. M. Shaheer Akhtar Chonbuk National University, Jeonbuk, South Korea shaheerakhtar@jbnu.ac.kr Scopus ID: 18933527100

Dr. L. N. Suvarapu University Environmental Engineering, Gyeongsan Republic of Korea suvarapu@gmail.com Scopus ID: 53982065500

Dr. Syed Adnan Ali Shah Faculty of Pharmcay, University Technology Universiti Malaysia Sarawak 94300 Kota MARA Cawangan Selangor Kampus Puncak Alam, 42300 Malavsia

benzene301@yahoo.com

Dr. Yodthong Baimark Department of Chemistry, Faculty of Science, Mahasarakham University Mahasarakham.

Thailand Yodthongb@gmail.com

Scopus ID: 22957332200

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Dr. Ravindra M. Kumbhare University of Bucharest Bucharest, Indian Institute of Chemical Technology, Hyderabad, India kumbhare@iict.res.in Istanciu75@yahoo.com Scopus ID: 7801611167

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CNR-Institute of Atmospheric Pollution

Métallurgie Appliquée, Tunisia

Mhadhbi_mohsen@yahoo.fr

Scopus ID: 24280679900

Dr. Attilio Naccarato

Research Rende (CS)

attilio.naccarato@iia.cnr.it

Scopus ID: 35726033800

Sasidhar.kantheti@gmail.com

Molecular Diagnostics and Therapeutics

Science, Kyung Hee University, Seoul,

Dept. of Chemical Biology Mount Sinai

Email: anuppandith@khu.ac.kr Scopus ID: 56736312500

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kmelody@unimas.my

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Dr. Vaishali Tomar

Scopus ID: 24461327700

Dr. Abdul Jabbar Al-Rajab

Center for Environmental Research

University of Massachusetts Medical

Worcester Woecester, 01655, USA

Jyothi.dhuguru@gmail.com Scopus ID: 47360948000

Hauppauge(NY)-11788, U.S.A

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Vaishali.aura08@gmail.com

Dr. Sripal Reddy Palavai

Carolina, 27606 U.S.A

Sripal27@gmail.com

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Samarahan.

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> kabobah@yahoo.com Scopus ID: 55246828400

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Dr. Hakan Arslan

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Department of Chemistry, University of Engineering and Management University of Delaware, Newark, DE, USA Kolkata, India Tanay. pramanik@gmail.com

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Dr. Gomaa Abdelgawad Mohammed Ali Dr. Malinee Sriariyanun Chemistry Department, Faculty of Science, King Mongkut's University of Technology Al-Azhar University, Assiut, 71524 Egypt gomaasanad@gmail.com Scopus ID: 38560963100

Prof. Dr. İlhan Uzun

Dicle University, Faculty of Education. Department of Chemistry Diyarbakir, 21280, United Arab Emirates Turkey iuzun21@yahoo.com

Dr. Naeem Uddin Siddigui M.J.P. Rohilkhand University Shahjahanpur, India naeemuddinsiddiqui@gmail.com

Dr. Mamedova Gunel

Department of the Azerbaijan National Academy of Sciences, Institute of natural resources, 76, Hevdar Alivev avenue. Nakhchivan, Azerbaijan gunelmamadova@mail.ru

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bkrishan.krg@gmail.com Scopus ID: 23002037000

Dr. Virendra Kumar Tiwari Department of Chemistry, University of Nebraska Lincoln 824-B Hamilton hall Lincoln, NE 685588-0304 USA virendrat21@gmail.com Scopus ID: 55790073900

Dr. Vandana Magarde Barkatullah University. Bhopal, India vmagarde@yahoo.co.in Scopus ID: 36114243100

Dr. Arshi Siddqui Barkatulla University Bhopal, India arshifazal@vmail.com

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Editorial Board

Dr. Mingiing Sun Department of Plant & Soil Sciences, smj@udel.edu Scopus ID: 57013900200

Dr. Mainak Banerjee

Birla Goa Campus, NH 17B, Bye Pass Road, Zuarinagar, 403726, Goa, India mainak@goa.bits-pilani.ac.in Scopus ID: 7201743845

North Bangkok 1518 Pracharat 1 Rd, Bangsue Bangkok, Thailand

macintous@gmail.com Scopus ID: 24473777500

Prof. Dinesh Chand Amity University Dubai Campus dsharma@amityuniversity ae Scopus ID: 36971665400

Dr. Bimal Krishna Banik

Department of Mathematics and Natural Science, Prince Mohammad Bin Fahd University, P. O. Box 1664; Al Khobar 31952, Kingdom of Saudi Arabia

University of Pitesti, Pitesti, Romania,

bimalbanik10@gmail.com Dr. Liviu Mitu

Romania University of Pitesti, Pitesti, Romania Ktm7ro@yahoo.com

Dr. Pham Tien Duc,

Faculty of Chemistry, VNU University of Science Vietnam National University, Hanoi Ciencias Aplicadas Ambientale(UDCA), 19 Le Thanh Tong, Hoan Kiem, Hanoi 100000. Vietnam

Dr. Haresh Thakellapalli

Department of Medicinal Chemistry, Pfizer Inc Insourcing-AviSyn Pharma Groton, CT, USA Haresh.thakellapalli@avisyn.com

Dr. Sibi Jose Barkatullah University, Bhopal, kojose@yahoo.com

Scopus ID: 49863564600

India

Editorial Assistants

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mohanmarugaraja@gmail.com Scopus ID: 16053259400

Dr. Julia Ubaskina

Laboratory of custom organic synthesis of Institute for Chemical Reagents and High Purity Chemical Substances of NRC, Moscow, 107076, Russia baseou@yandex.ru Scopus ID: 57192817935

Dr. Tawkir Sheikh

Department of Chemistry, BITS-Pilani- K. K. Arts, Com. and Science College, Tukum Chandrapur- 442401 India msheikh196@gmail.com

Scopus ID: 35318827700

Dr Amrita Chatterjee

Department of Chemistry, BITS Pilani KK Birla Goa Campus, NH 17 B Zuarinagar, Goa. 403726. India amrita@goa.bits-pilani.ac.in

Dr. Nelson R. Villarante

Department of Physical and Sciences College of Arts and Sciences, University of the Philippines Manila nrvillarante@up.edu.ph Scopus ID: 15047313200

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Department of Experimental and Clinical Pharmacology University of Rzeszów, Poland karolw222@wp.pl

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